



# **TRINITY COLLEGE FOR WOMEN NAMAKKAL**

## **Department of Physics**

### **Statistical Mechanics**

### **23PPH04-EVEN Semester**

**Presented by**

**Dr.R.SAKUNTHALADEVI**

**Assistant Professor**

**Department of Physics**

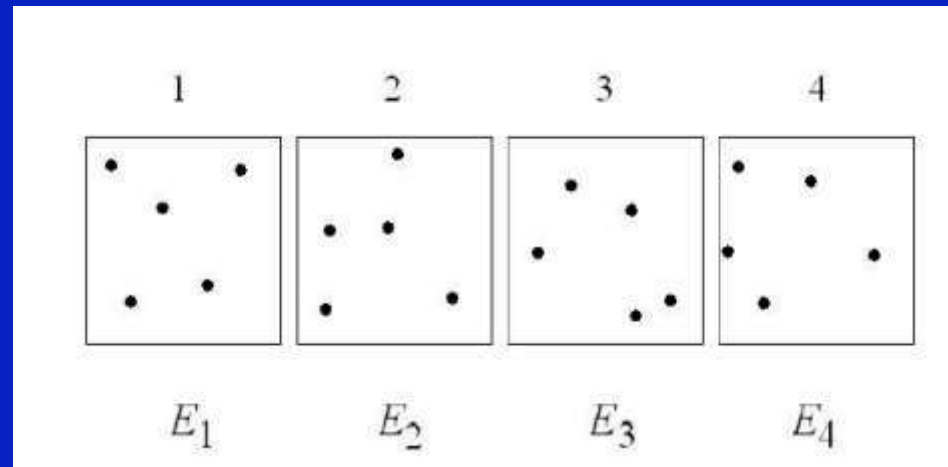
**<http://www.trinitycollegenkl.edu.in/>**

# Introduction

- ❖ Chemical Sciences and Statistical Physics are a collection of a number of macroscopically identical but essentially independent systems.
- ❖ The System, is defined by the collection of a large number of particles
- ❖ Same macroscopic conditions, like Volume, Energy, Pressure, Temperature and the total number of particles.

# Microstates and Ensembles:

- ❖ According to the quantum mechanics at any given moment, the system is in a superposition of distinct quantum states called microstates.
- ❖ An ensemble on the other hand is the collection of all possible microstates consistent with the externally controlled macroscopic parameter.



# Fundamental postulate

- ❖ The probability density function is proportional to some function of the ensemble parameters and random variables.
- ❖ Thermodynamic state functions are described by ensemble averages of random variables.
- ❖ The entropy as defined by Gibbs's entropy formula matches with the entropy as defined in classical thermodynamics

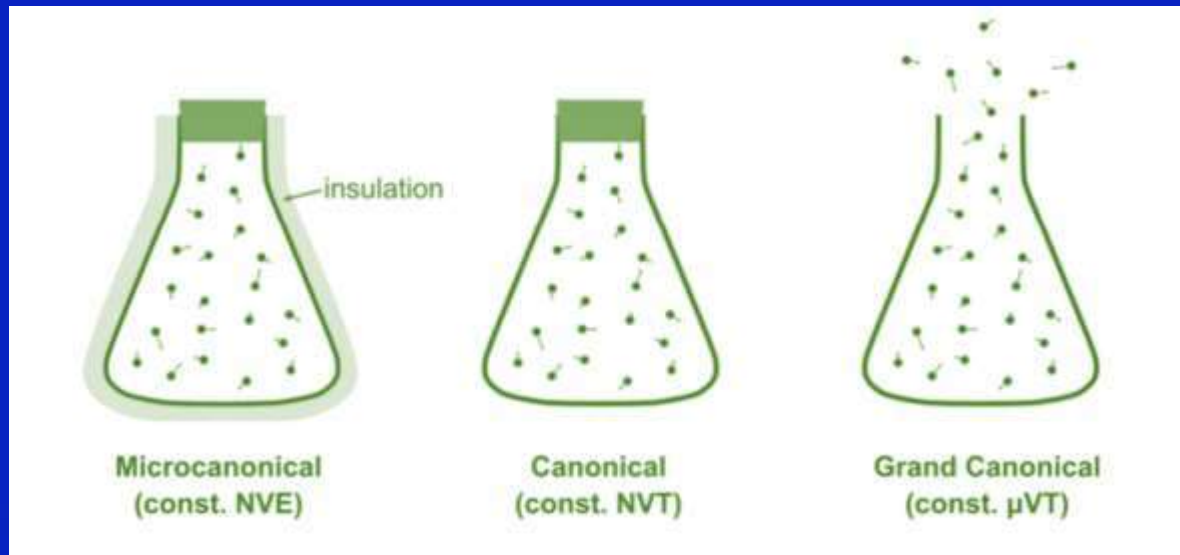
# Types of Ensembles

*There are three types of ensembles:*

Micro-canonical Ensemble

Canonical Ensemble

Grand Canonical Ensemble



# Micro-canonical Ensemble

- ❖ It is the collection of a large number of essentially independent systems having the same energy  $E$ , volume  $V$  and total number of particles  $N$ .
- ❖ Separated by rigid impermeable and insulated walls, such that the values of  $E$ ,  $V$  &  $N$  are not affected by the mutual pressure of other systems.

# Canonical Ensemble

- ❖ It's the collection of a large number of essentially independent systems having the same temperature  $T$ , volume  $V$  and the number of particles  $N$ .
- ❖ The equality of temperature of all the systems can be achieved by bringing all the systems in thermal contact.
- ❖ Hence, in this ensemble, the systems are separated by rigid, impermeable but conducting walls, the outer walls of the ensemble are perfectly insulated and impermeable though.

# Grand Canonical Ensemble

- ❖ It is the collection of a large number of essentially independent systems having the same temperature  $T$ , volume  $V$  & chemical potential  $\mu$ .
- ❖ The systems of a grand canonical ensemble are separated by rigid permeable and conducting walls.



# Ensemble Average

- ❖ Every statistical quantity has not an exact but an approximate value.
- ❖ The average of a statistical quantity during motion is equal to its ensemble average.
- ❖ Let  $R(x)$  be a statistical quantity along the  $x$ -axis and  $N(x)$  be the number of phase points in phase space, then **the ensemble average** of the statistical quantity  $R$  is defined as,

$$\bar{R} := \frac{\int_{-\infty}^{\infty} R(x)N(x)dx}{\int_{-\infty}^{\infty} N(x)dx}$$

# THANK YOU

<http://www.trinitycollegenkl.edu.in/>